

## Ni-Cr-Al Alloy for Neutron Scattering at High Pressures

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### Abstract

High-pressure devices used in neutron scattering require materials with high strength, low neutron attenuation, and no activation. They should also be non-magnetic at all temperatures. Only very few materials fulfill these criteria. Here, we report on the manufacture and properties of the Ni-Cr-Al alloy (57.0 wt.% Ni, 40.0 wt.% Cr, and 3.0 wt.% Al) that could be used in diamond anvil and clamp cells. The casting and heat treatment to obtain the material with the yield strength of about 1.7 GPa are described. It has no magnetic order down to at least 3.7 K as evidenced by diffraction with polarized neutrons. Synchrotron powder diffraction reveals that phases of the Ni<sub>2</sub>Cr, Cr, and Ni<sub>3</sub>Al types are present.

**Keywords:** neutron scattering, high pressure, magnetism, mechanical properties, Ni-Cr-Al alloy

## Introduction

Materials that could be used in high-pressure devices for neutron scattering require not only high strength but also no neutron activation. Their neutron attenuation is to be as low as possible. They should also be non-magnetic at very low temperatures. Among them, the Ni-Cr-Al alloy (NiCrAl), frequently called the *Russian alloy*, is considered to be the best [1-6]. It consists of only three elements (57.0 wt.% Ni, 40.0 wt.% Cr, and 3.0 wt.% Al) and is hardened by precipitation. According to the Ni-Cr phase diagram [7,8], Cr dissolves completely in Ni (fcc) at high temperatures for this composition and forms a supersaturated solid solution (Ni, Cr) after quenching. A precipitate of Ni<sub>2</sub>Cr and a Ni-bearing Cr forms during annealing at temperatures above 500 °C. Al is a precipitation former in nickel-based alloys, which has a strengthening effect.

At room temperature, NiCrAl has a yield strength of about 2.00 GPa, a tensile strength of 2.10 GPa, and an elongation of about 7% [1]. It is possible to reach an ultimate tensile strength of up to 2.37 GPa, but with an elongation of less than 3% [2]. Eremets [1] has indicated that high-pressure cells could be made of this material when its hardness on the Rockwell C scale (HRC) is in the range 52-56, while the corresponding yield strength is 1.65-1.75 GPa. The material with higher hardness is too brittle. According to his data, the yield strength ( $\sigma_{0.2}$ ) is linearly correlated with the HRC:  $\sigma_{0.2} = 0.0264 \times \text{HRC} + 0.2636$ . Moreover, small temperature dependence of resistivity and magnetic susceptibility of NiCrAl is beneficial for its applications in neutron scattering [6,9].

NiCrAl has been used in high-pressure technology (especially for clamp cells) in the former Soviet Union for many decades [1,6] and is not available outside the Russian Federation. Uwatoko et al. [9] made an attempt to produce a small quantity of this material containing boron. The Ni-Cr-Al alloy was annealed at 1200 °C, followed by water quenching. It was then

aged between 400 °C and 900 °C. There are two important issues that Uwatoko et al. [9] did not clarify in their article. They added boron to NiCrAl but they did not indicate which isotope it was. If the added boron is not pure  $^{11}\text{B}$ , this material cannot be effectively used for construction of low-attenuation pressure cells for neutron scattering. They also did not mention whether there was any oxidation of the material during the heat treatment and, especially, after water quenching.

Recently, we have started developing various pressure cells for neutron scattering and magnetic measurements. Currently, our diamond anvil cells are made of the Cu-Be alloy [10]. To improve their performance upon compression, it would be beneficial to use materials with better mechanical properties like, for instance, NiCrAl. Since the *Russian alloy* is not available to us, we have attempted to manufacture it ourselves. Here, we report on our production process and characterization of NiCrAl using a suite of various experimental techniques.

## Methods

The *Russian alloy* was prepared with the composition 57.0 wt.% Ni, 40.0 wt.% Cr and 3.0 wt.% Al to obtain the material with a weight of 50 kg. The Ni pellets (99.95%), Cr granules (99.95%), and Al granules (99.99%) were melted in an  $\text{Al}_2\text{O}_3$ -CaO spinel crucible under an Ar atmosphere (700 mbar of pressure) at 1500°C. Subsequently, the furnace (PVA VSG 100) was switched off and cooled down to room temperature in the Ar atmosphere. The casted rod was about 80 cm in length and 11 cm in diameter.

Several small samples were cut from the rod using spark erosion. Their heat treatment was performed under an Ar flow. The furnace was heated with the rate 300 °C/h to various temperatures. After dwelling for 2 hours at the highest temperature, the furnace was shut down and the Ar flow was increased to improve the cooling.

The surface of all samples after the heat treatment was diamond polished to make it smooth and to remove any oxidation products. The Vickers hardness measurements were carried out using the indenter Durimet (Leitz). Nanoindentation tests were performed on a Nanomechanics iNano instrument using a diamond Berkovich indenter calibrated prior to indentation via multiple indents into fused silica [11]. An array of 10×10 indents with the nearest-neighbour distance of 50 μm was made for the measurements on each sample. The unit of the obtained hardness is in GPa and is converted to the Vickers scale (HV) using the relation  $HV = GPa/9807$ . The HV→HRC conversion formula is:  $HRC = (100 \times HV - 14500)/(HV + 223)$ .

Thin samples of the as-casted (raw) material (0.16 mm in thickness) and of the aged material at 750 °C (0.11 mm in thickness) were investigated with powder diffraction on the beamline P24 at the synchrotron Petra-III in Hamburg using a marCCD165 detector calibrated with a Si standard ( $\lambda = 0.494 \text{ \AA}$ ). The two-dimensional images were integrated with the program FIT2D to yield intensity versus  $2\theta$  diagrams [12]. Le Bail refinements of the diagrams were carried out with the program Jana2006 [13].

Small pieces of the raw sample as well as of the one aged at 750 °C (44 mg and 34 mg, respectively) were studied on the Quantum Design System *Dynacool* with the vibrating sample magnetometer option (VSM). Two measurements were made on each sample. One was to measure the magnetic moment at 0.1 T from 300 K down to 1.9 K with the sweep rate of 2 K/min (the M-T curve). Another one was at 2 K from 7 T to -7 T then back to 7 T, with the sweep rate of 100 Oe/sec (the M-H curve).

Polarized neutron diffraction investigations were performed at the cold-neutron spectrometer DNS ( $\lambda = 4.2 \text{ \AA}$ ) in the Heinz Maier-Leibnitz Zentrum (MLZ) in Garching. The neutron transmission measurement was carried out at the neutron reflectometer TREFF ( $\lambda = 4.74 \text{ \AA}$ ) in MLZ (Garching).

## Results and Discussion

Initially, a heat treatment process similar to that mentioned by Uwatoko et al. [9] was performed, i.e., a small piece was heated at 1200 °C for one hour in the air and then water quenched. However, severe oxidation of the sample can be seen afterwards, which is a green layer of Cr<sub>2</sub>O<sub>3</sub>. Uwatoko et al. [9] did not mention whether there was any oxidation of their material and whether their heat treatment processes were indeed in the air or in an inert atmosphere. Water quenching is necessarily performed in the air, though. To gain a better understanding of the annealing process, another sample (untreated at 1200 °C) and the water-quenched one were both aged at 750 °C. The hardness of the sample, measured with the Vickers method, without any heat treatment is 54 HRC. Its hardness decreases to 13 HRC after the annealing at 1200 °C and water quenching. It then increases to 51 HRC after aging at 750 °C. It is likely that the oxidation influences the hardness of NiCrAl. The hardness of the piece with no annealing at 1200 °C and no water quenching increases to 57 HRC after aging at 750 °C. On the basis of these hardness tests, a simple aging process without the annealing at 1200 °C and water quenching was subsequently used to further characterize the alloy.

The dimensions of the samples changed after the heat treatment. The sample after annealing at 1200 °C and water quenching was 0.3-0.8% larger than before. The same sample after additional annealing at 750 °C is similar in size to the raw material. The sample after aging at 750 °C only is 0.1-0.25% smaller than before the heat treatment.

Four different samples of the as-casted *Russian alloy* (untreated at 1200°C) were aged at 500 °C, 600 °C, 700 °C, and 750 °C, respectively, for two hours under the constant Ar flow. All the samples were slightly coloured. The colour change can be correlated with the annealing temperature. The higher the temperature, the darker blue colour can be seen. The average hardness of our samples aged at different temperatures is shown in **Figure 1**. The tests were

performed using both the Vickers and nanoindentation methods. As expected, the hardness of our aged alloy is higher than that of the raw sample without any heat treatment. Although, it is slightly lower than that determined on the NiCrAl produced by Uwatoko et al. [9], our data correlate well with the data from other laboratories. Altogether, the hardness of all our samples is within the range HRC = 52-56 that is optimal for manufacturing high-pressure cells [1].

Synchrotron powder diffraction diagrams of two samples produced in this study are shown in **Figure 2**. They are very similar to each other indicating that aging does not have any substantial effect on the phase composition of the NiCrAl alloy. Phases of the Ni<sub>2</sub>Cr (Immm, Z = 2), Cr (Im-3m, Z = 2), and Ni<sub>3</sub>Al (Pm-3m, Z = 1) types can be identified (**Table 1**). The presence of the Cr-type phase is documented as a poorly resolved shoulder at about  $2\theta = 13.8^\circ$ . The lattice of the orthorhombic Ni<sub>2</sub>Cr phase is derived from the fcc one:  $a \approx a_{fcc}/\sqrt{2}$ ,  $b \approx 3a_{fcc}/\sqrt{2}$ , and  $c \approx a_{fcc}$ . Altogether, aging has hardly any effect on the lattice parameters of all three phases.

The M-T and M-H curves are shown in **Figure 3**. For the sample aged at 750°C, the magnetization (M) at 1.9 K and 300 K is 0.013 emu/g and 0.005 emu/g, respectively. For the sample without any aging, the magnetization is 0.0096 emu/g and 0.0035 emu/g at 1.9 K and 300 K, respectively. This is consistent with the results from Sadykov (2016), which show that the magnetization is 0.013 emu/g at very low temperatures and 0.0035 emu/g at 300 K. It can be seen in the M-T curves that the magnetization is nearly constant from 300 K to 25 K (**Figure 3**). It then suddenly increases at lower temperatures. The magnetization is zero when there is no external field. It means that the material is paramagnetic.

Neutron transmission of the raw sample (the thickness of 6 mm, the density of 7.6 g/cm<sup>3</sup>) measured using the neutron reflectometer TREFF at MLZ (Garching) is 47.5%. Assuming the attenuation formula as  $I/I_0 = \exp(-\mu*d)$ , the transmission of the sample with the same thickness but made of the CuBe alloy could be compared to this value. The attenuation coefficient of CuBe is wavelength dependent:  $\mu_{CuBe} = 0.1611\lambda + 0.041$  (cm<sup>-1</sup>) [14]. Accordingly, the sample

made of CuBe would have the transmission of 61.7% at the wavelength of 4.74 Å. Hence, the *Russian alloy* transmits fewer neutrons than CuBe. On the other hand, its mechanical and magnetic properties are far superior to those of CuBe.

Diffraction measurements on the heat untreated sample using polarized neutrons were performed at the beamline DNS (MLZ, Garching). The results show (**Figure 4**) that the NiCrAl alloy is magnetically disordered even at 3.7 K. Two nuclear coherent scattering peaks at  $1.75 \text{ \AA}^{-1}$  and  $2.49 \text{ \AA}^{-1}$  (d-spacing of 3.59 Å and 2.52 Å, respectively) correspond to the first two Bragg (001) and (101) reflections of the Ni<sub>3</sub>Al type phase in synchrotron powder diffraction diagrams shown in **Figure 2**.

## Conclusions

We have successfully manufactured the Ni-Cr-Al alloy with the yield strength of about 1.7 GPa that is optimal for neutron scattering at high pressures. The material has no magnetic order down to at least 3.7 K. It could be used in various devices including diamond anvil and clamp cells.

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**Table 1** Lattice parameters of the phases observed in the synchrotron powder diffraction patterns of the sample with no aging and of the one aged at 750°C.

	$a$ (Å)	$b$ (Å)	$c$ (Å)
<i>No aging</i>			
Ni <sub>2</sub> Cr type	2.5523(5)	7.602(1)	3.5644(5)
Cr type	2.8801(2)		
Ni <sub>3</sub> Al type	3.5811(2)		
<i>750°C</i>			
Ni <sub>2</sub> Cr type	2.5559(3)	7.578(1)	3.5666(4)
Cr type	2.8808(2)		
Ni <sub>3</sub> Al type	3.5773(1)		

## Figure captions

**Figure 1** Hardness of the samples aged at different temperatures in this study (yellow diamonds and blue triangles) compared to the hardness of the alloy by Uwatoko et al. [9] determined at other laboratories.

**Figure 2** Synchrotron powder diffraction patterns of the sample without any heat treatment and the one aged at 750°C (black curves). The ticks indicate the positions of the Bragg reflections due to the Ni<sub>2</sub>Cr, Cr, and Ni<sub>3</sub>Al type phases (the top, middle, and bottom rows, respectively). The (001) and (101) reflections of the Ni<sub>3</sub>Al-type phase are indicated. The difference between the measured and calculated lines using the Le Bail method, assuming the presence of the three phases, is drawn as a blue line for each sample.

**Figure 3** M-T and M-H curves for the sample aged at 750°C (top) and for the sample without any heat treatment (bottom).

**Figure 4** Nuclear coherent (black symbols), magnetic (green symbols), and spin incoherent (red symbols) contributions to scattering of the heat untreated sample at 100 K (left) and 3.7 K (right) measured at the instrument DNS (MLZ, Garching).

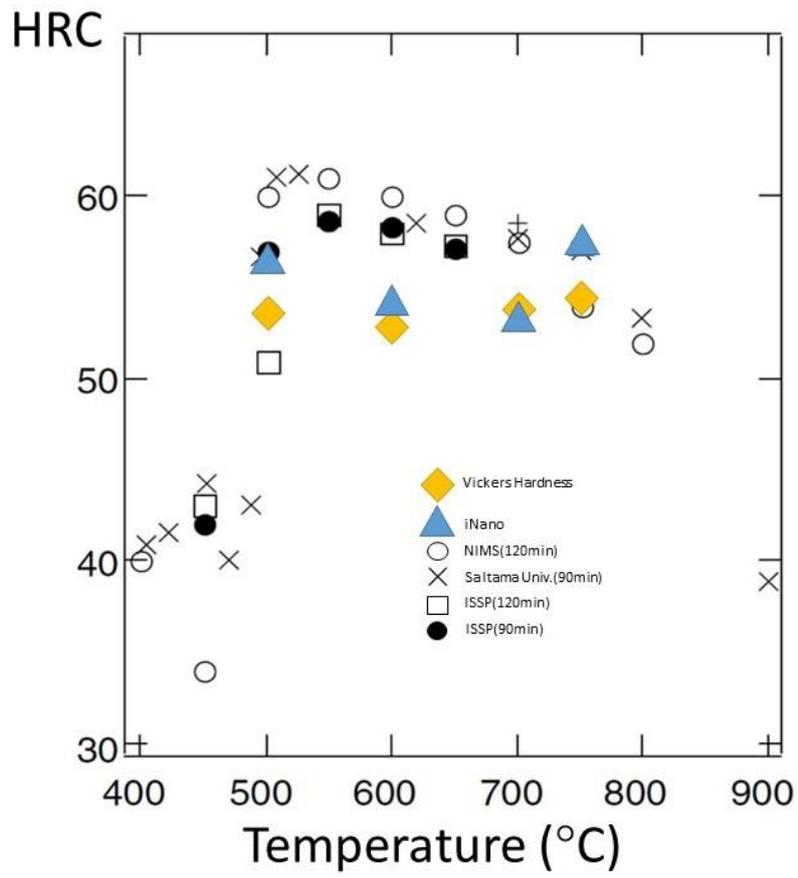


Figure 1.

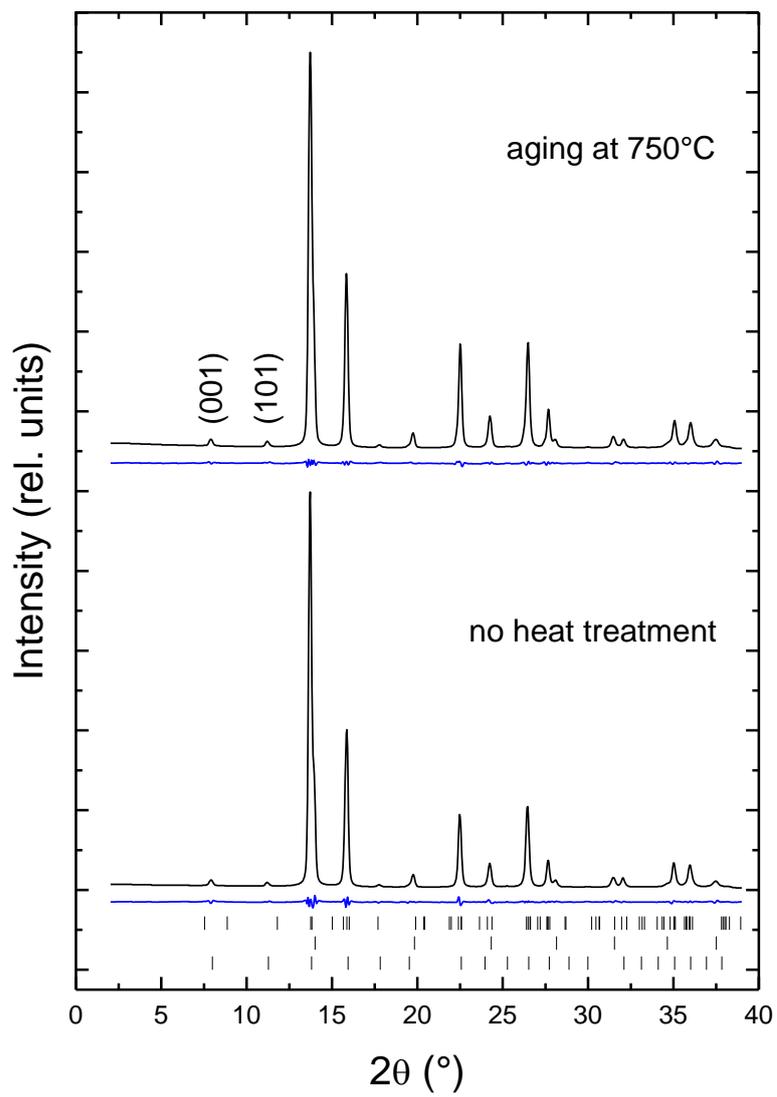
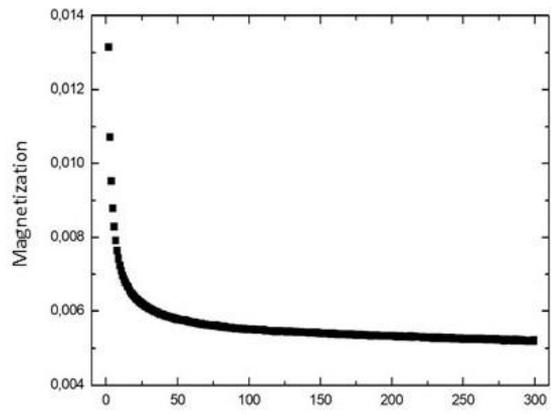
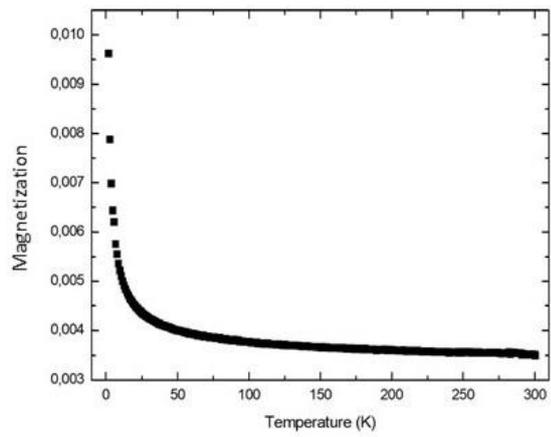
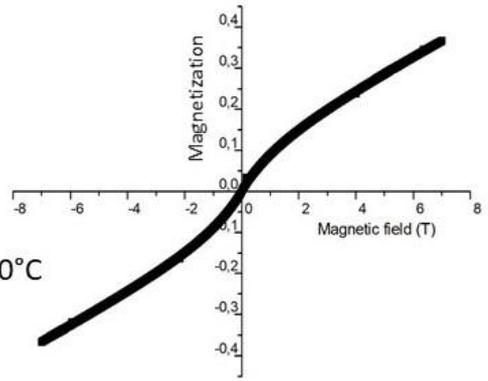


Figure 2.



aged at 750°C



no heat treatment

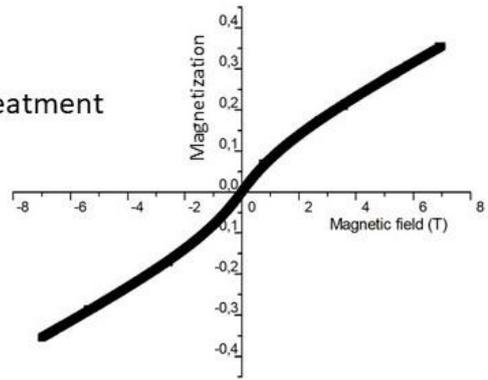


Figure 3.

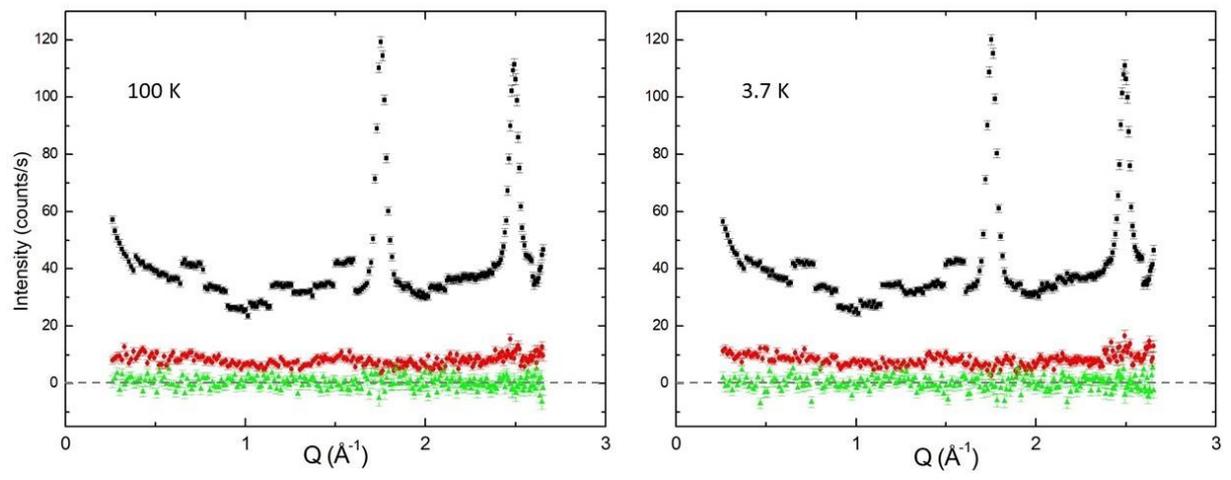


Figure 3.